

# A parametric Boltzmann equation analysis of CO<sub>2</sub> dissociation in cold plasmas and afterglows.

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**PROGRESS IN THE NON-EQUILIBRIUM VIBRATIONAL KINETICS  
OF HYDROGEN IN MAGNETIC MULTICUSP H<sup>-</sup> ION SOURCES**

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**From dynamics to modeling of plasma complex systems:  
negative ion (H<sup>-</sup>) sources**

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**Deactivation dynamics of vibrationally excited nitrogen molecules  
by nitrogen atoms.**

**Effects on non-equilibrium vibrational distribution  
and dissociation rates of nitrogen under electrical discharges**

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**CHEMICAL  
PHYSICS  
LETTERS**

**The effect of N+N<sub>2</sub> collisions  
on the non-equilibrium vibrational distributions of nitrogen  
under reentry conditions**

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# Outline

- Main idea: role of CO<sub>2</sub> vibrational excitation in dissociation
- Dissociation by Pure Vibrational Mechanisms (PVM)
- Dissociation by Direct Electron Impact (DEI)
- Electron Boltzmann equation
- Continuous Discharges
- Post-discharge conditions
- Conclusions

# Main idea:

Increase the dissociation of CO<sub>2</sub> by using the vibrational energy introduced by electron molecule collisions followed by vibrational climbing up to the dissociation limit.

The necessary power is that one to excite the vibrational levels of the CO<sub>2</sub> molecule much lower than that one requested for thermal dissociation of CO<sub>2</sub> as well as for electron impact dissociation.

A. Fridman et al. (1977)<sup>1,2</sup>  
E. Molinari et al. (1976)<sup>3,4</sup>

A. Bogaerts et al. (2014)<sup>5,6</sup>  
T. Silva et al. (2014)<sup>7</sup>  
R. Van de Sanden et al. (2014)<sup>8</sup>

<sup>1</sup> A. Fridman, *Plasma Chemistry*, Cambridge University Press, New York (2008)

<sup>2</sup> V. A. Legasov, V. K. Givotov, E. G. Krashennikov, V. D. Rusanov, A. Fridman, *Sov. Phys. Doklady* **238**, 66 (1977)

<sup>3</sup> P. Capezzutto, F. Cramarossa, R. D'Agostino and E. Molinari, *J. Phys. Chem.* **80**, 882 (1976)

<sup>4</sup> M. Capitelli and E. Molinari, *Topics in Current Chemistry* **90**, 59 (1980)

<sup>5</sup> T. Kozak and A. Bogaerts, *Plasma Sources Sci. and Technol.* **23**, 045004 (2014)

<sup>6</sup> T. Kozak and A. Bogaerts, *Plasma Sources Sci. and Technol.* **24**, 015024 (2015)

<sup>7</sup> T. Silva, N. Britun, T. Godfroid and R. Snyders., *Plasma Sources Sci. and Technol.* **23**, 025009 (2014)

<sup>8</sup> A. Goede et al., XARMAE Workshop, Barcelona (2014)

# Dissociation by Pure Vibrational Mechanism (PVM)

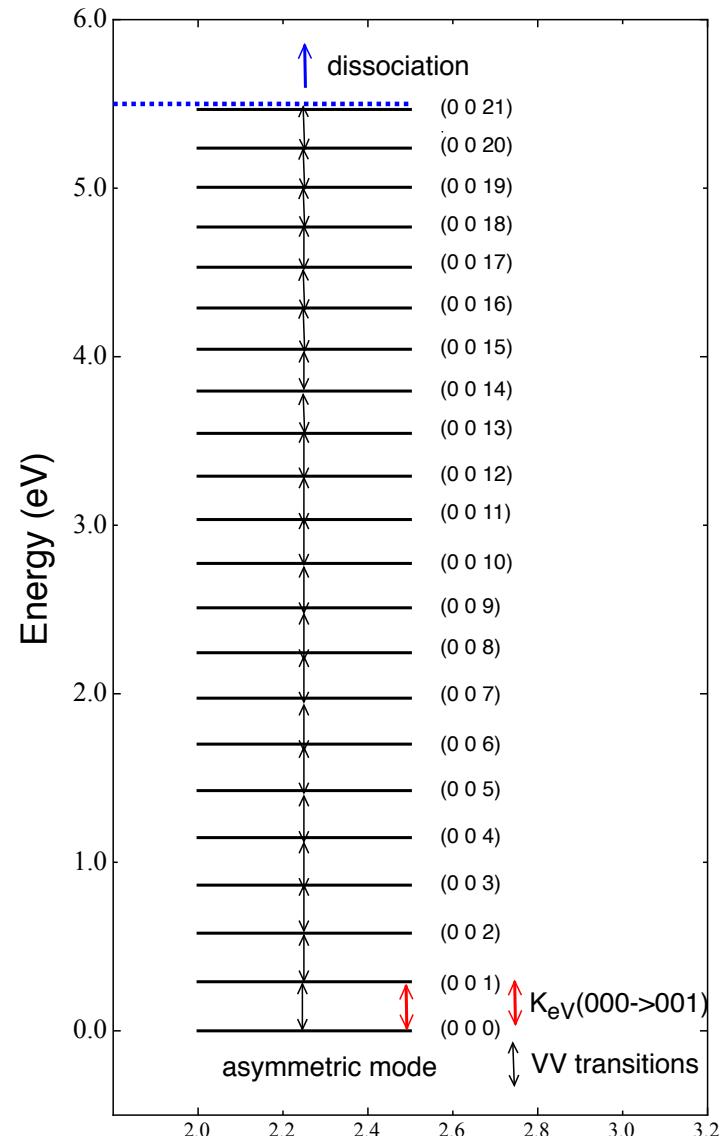
- **eV (electron vibration)** energy exchange processes;
- **VV (vibration vibration)** and **VT (vibrational translation)** energy exchange processes;
- **Dissociation** linking the last vibrational level and the continuum.

PVM Upper Limit dissociation rates

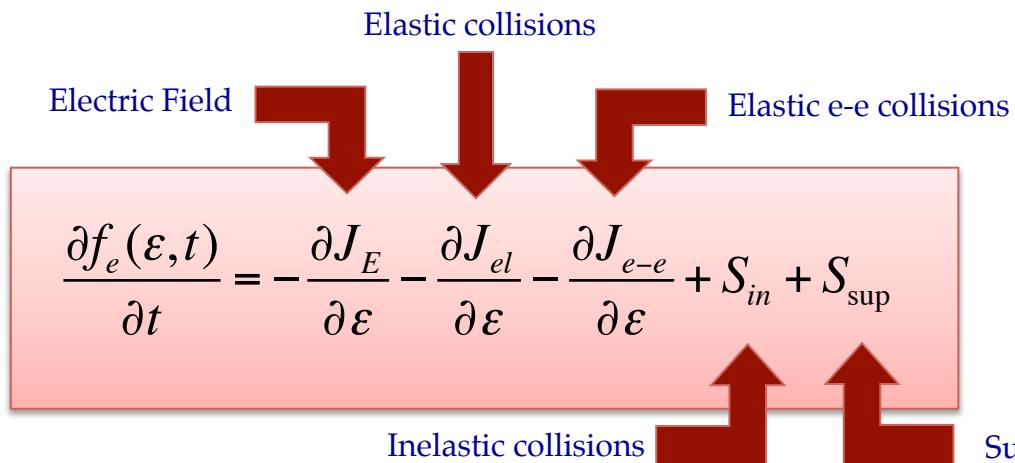
$$K_d^{(ulPVM)} = \frac{1}{v_{\max}} k_{eV} (000 \rightarrow 001)$$

including the effect of excited vibrational levels

$$K_d^{(ulPVM)}(all) = \frac{1}{v_{\max}} \sum_n \frac{\epsilon_{v_n}}{\epsilon_{v_8}} k_{eV}(v_0 \rightarrow v_n)$$



# Electron Boltzmann Equation (BE)



## Parametric solution of the BE

- E/N
- $T_1$  and  $T_2$  vibrational temperatures  
(Two temperature Boltzmann distrib.)
- $\chi_e$  electron molar fraction
- $\text{CO}_2(e_2)$  electr. excited state population

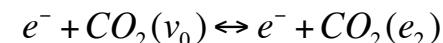
## Phelphs database

$\text{CO}_2(v_0)$	(000)	0.00
$\text{CO}_2(v_1)$	(010)	0.083
$\text{CO}_2(v_2)$	(020)+(100)	0.167
$\text{CO}_2(v_3)$	(030)+(110)	0.252
$\text{CO}_2(v_4)$	(0n0)+(n00)	0.339
$\text{CO}_2(v_5)$	(0n0)+(n00)	0.442
$\text{CO}_2(v_6)$	(0n0)+(n00)	0.505
$\text{CO}_2(v_7)$	(0n0)+(n00)	2.5
$\text{CO}_2(v_8)$	(001)	0.291
$\text{CO}_2(e_1)$		7.0
$\text{CO}_2(e_2)$		10.5
$\text{CO}_2^+$		13.3

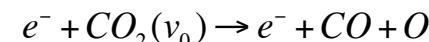
### Vibrational Excitation/De-excitation



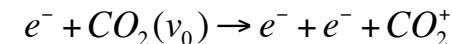
### Electronic Excitation/De-excitation (10.5 eV)



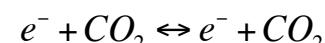
### Dissociative Excitation (7.0 eV)



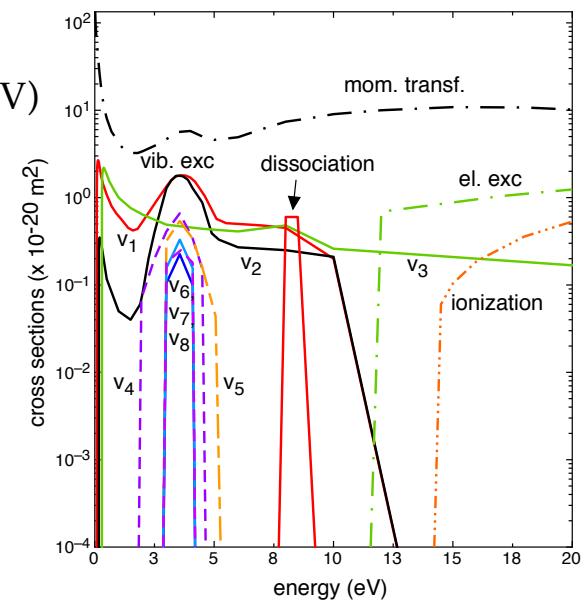
### Ionization



### Momentum transfer



## E-impact cross sections



\* A. V. Phelphs Database: <http://www.lxcat.laplace.univ-tlse.fr>

# Direct electron impact dissociation

$$k_d(000) = \int_{E_{thr}} f(\varepsilon) \sigma(\varepsilon) v(\varepsilon) d\varepsilon$$

Including the effect of excited vibrational levels

## Assumptions

- Shift of cross section thresholds (->  $\exp(1/T_e)$ )
- Boltzmann population for the excited states at  $T_v$

$$K_d(all) = \sum_v^{v_{\max}} \exp\left[\frac{\varepsilon_{00v}}{k_B} \left(\frac{1}{T_e} - \frac{1}{T_v}\right)\right] k_d(000)$$

For  $T_e = T_v$        $K_d(all) = v_{\max} k_d(000)$

# CO<sub>2</sub> Continuous Discharge: **Stationary EEDF**

E/N=15-30-50-80 Td

T<sub>1</sub> symmetric, bending levels

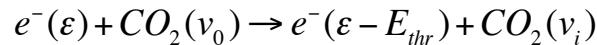
T<sub>2</sub> first asymmetric level (001), CO<sub>2</sub>(e<sub>2</sub>)

no e-e

Cold case

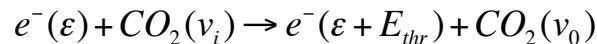
$$T_1 = 0, T_2 = 0$$

inelastic collisions

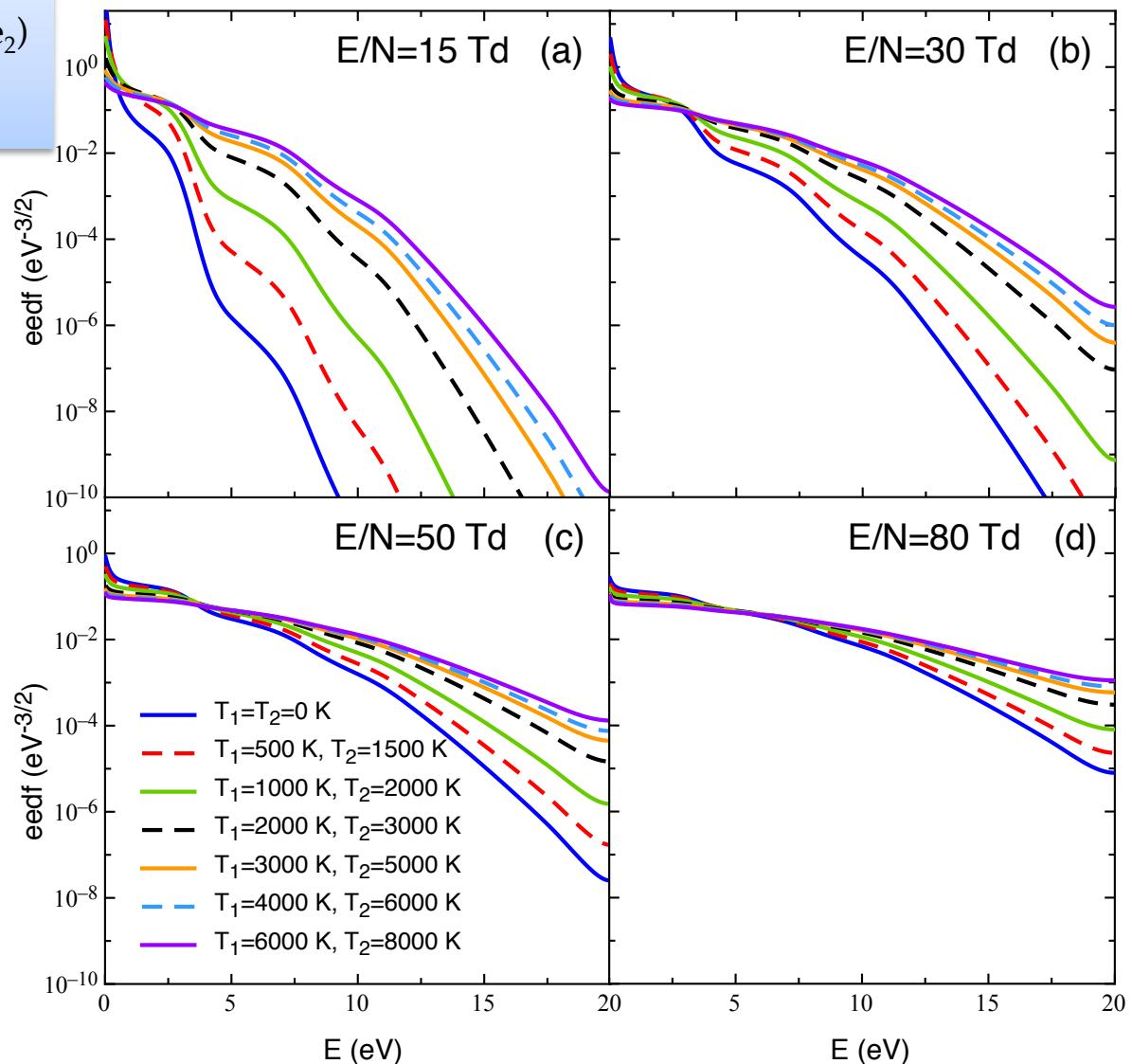


$$T_1 \neq 0, T_2 \neq 0$$

superelastic vibr. and electr. collisions

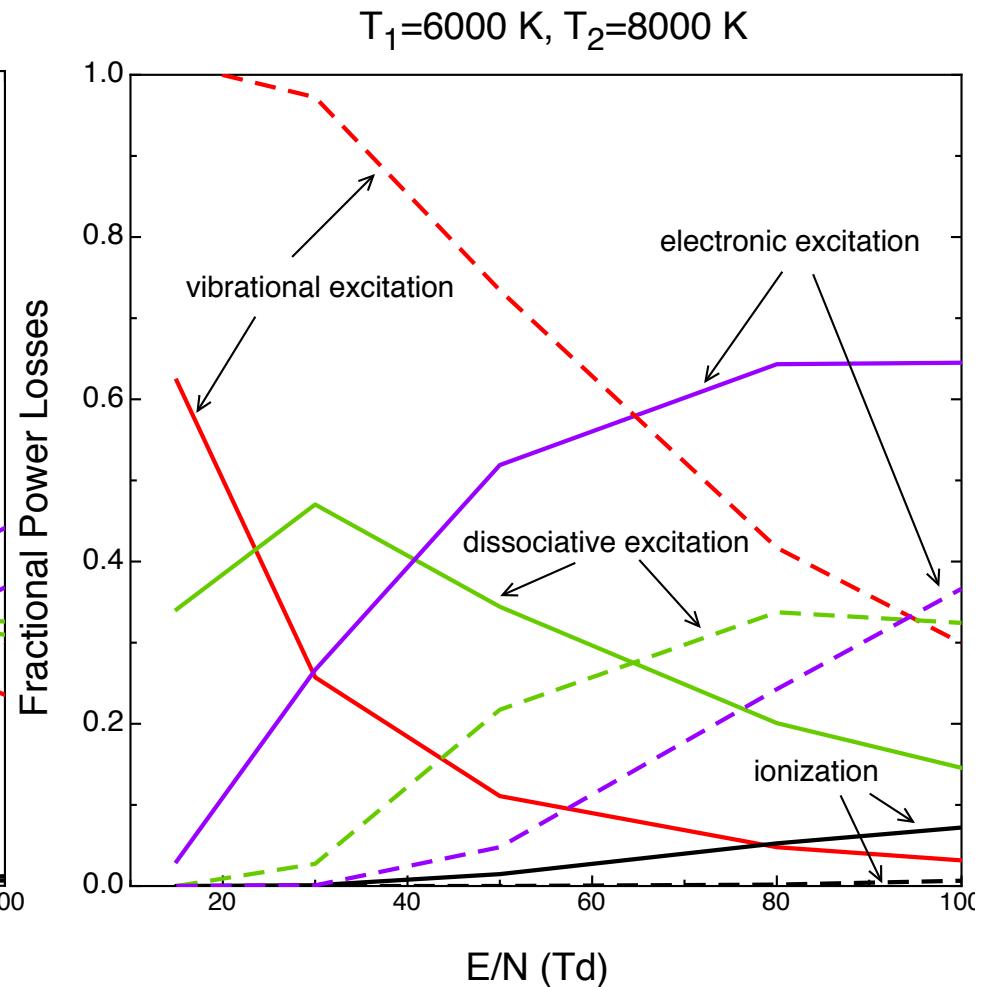
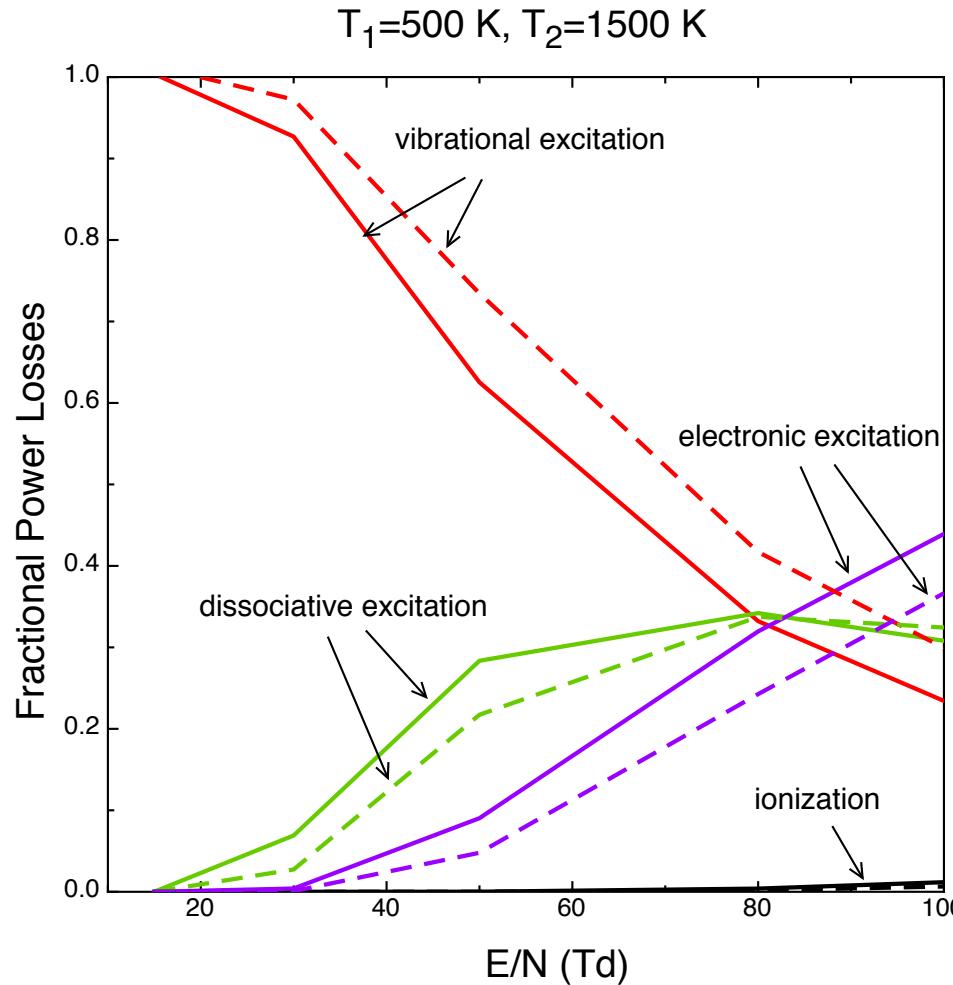


**Electron Energy Distribution Function (eedf)**



# CO<sub>2</sub> Continuous Discharge: Fractional Power Losses

Cold gas approximation (dashed lines)  
with (full lines) and without (dashed lines) **superelastic collisions**



# CO<sub>2</sub> Continuous Discharge: Dissociation Rates

Direct electron impact

$$k_d(000) = \int_{E_{thr}} f(\varepsilon) \sigma(\varepsilon) v(\varepsilon) d\varepsilon$$

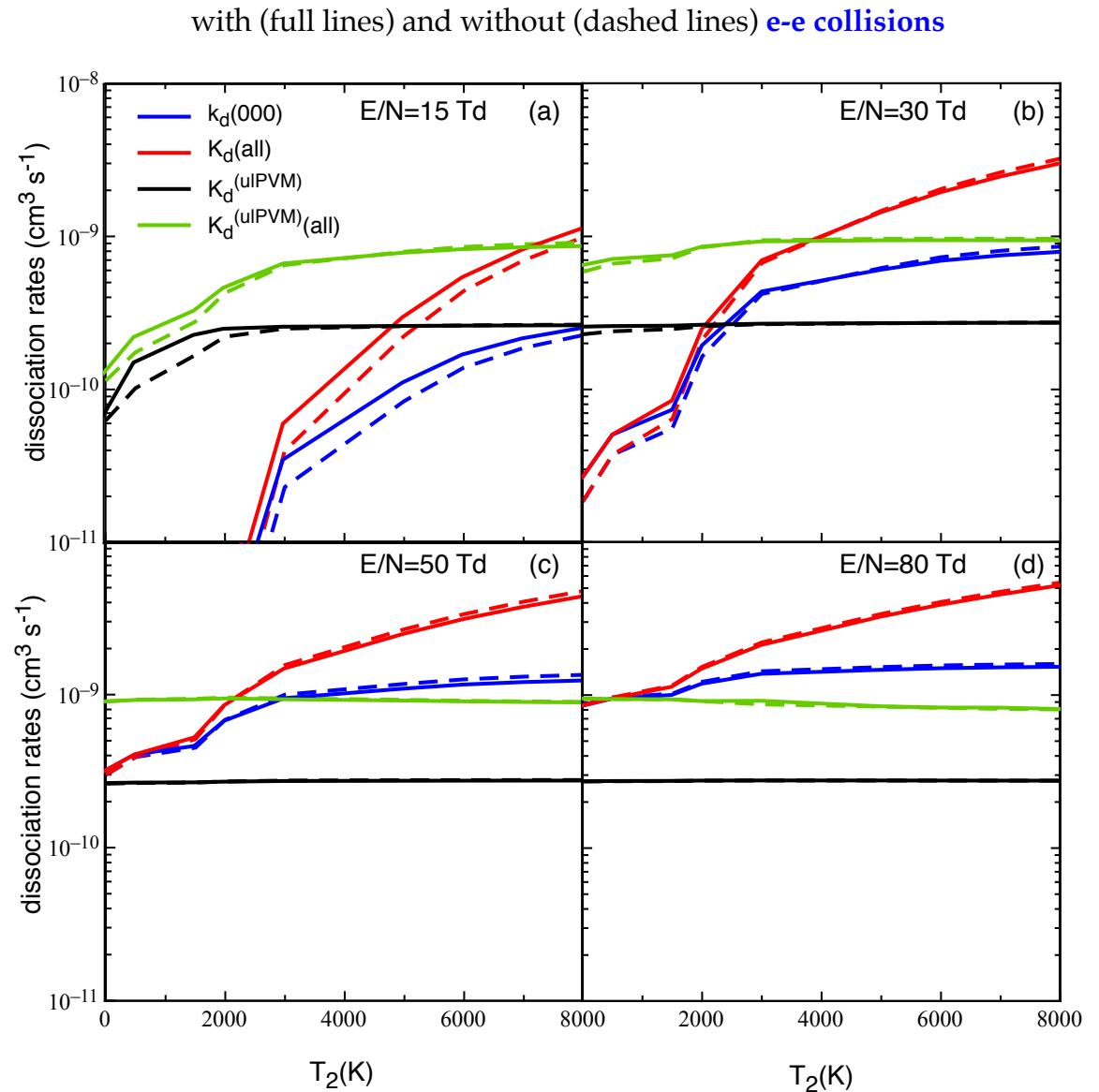
$$K_d(all) = \sum_v^{\nu_{\max}} \exp\left[\frac{\varepsilon_{00v}}{k_B} \left(\frac{1}{T_e} - \frac{1}{T_v}\right)\right] k_d(000)$$

Pure Vibrational Mechanism

$$K_d^{up-PVM}(001) = \frac{1}{\nu_{\max}} K_{eV}(000 \rightarrow 001)$$

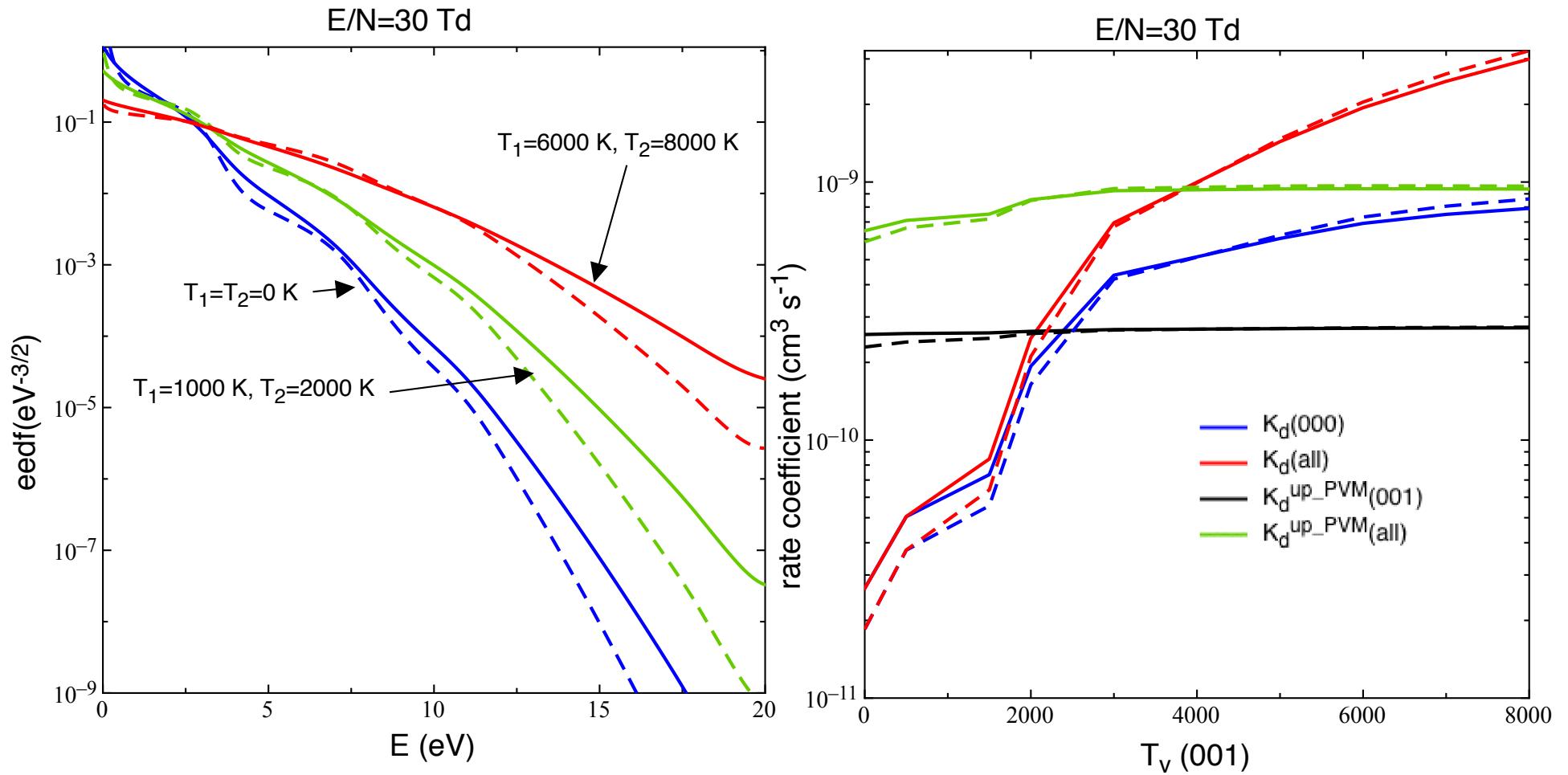
$$K_d^{up-PVM}(all) = \frac{1}{\nu_{\max}} \sum_{n=1}^8 \frac{\varepsilon_{\nu_n}}{\varepsilon_{\nu_8}} K_{eV}(0 \rightarrow n)$$

This results are important for guiding the different experiments especially for the **MW sustained plasma**.



# CO<sub>2</sub> Continuous Discharge: **e-e collision effect**

with (full lines) and without (dashed lines) **e-e collisions**

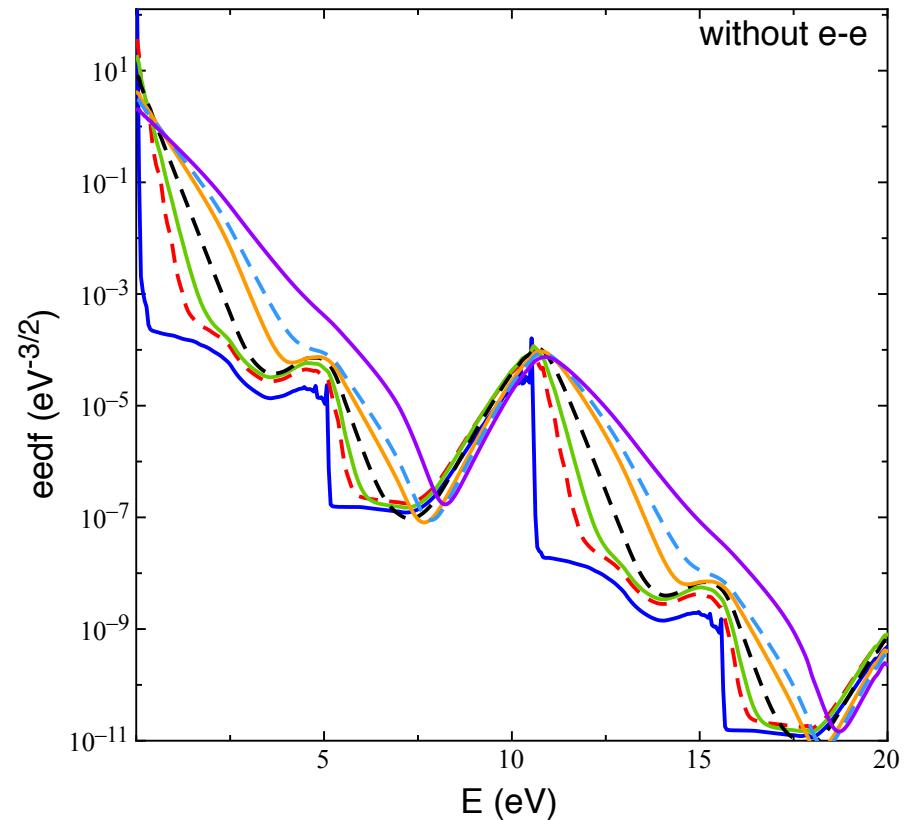
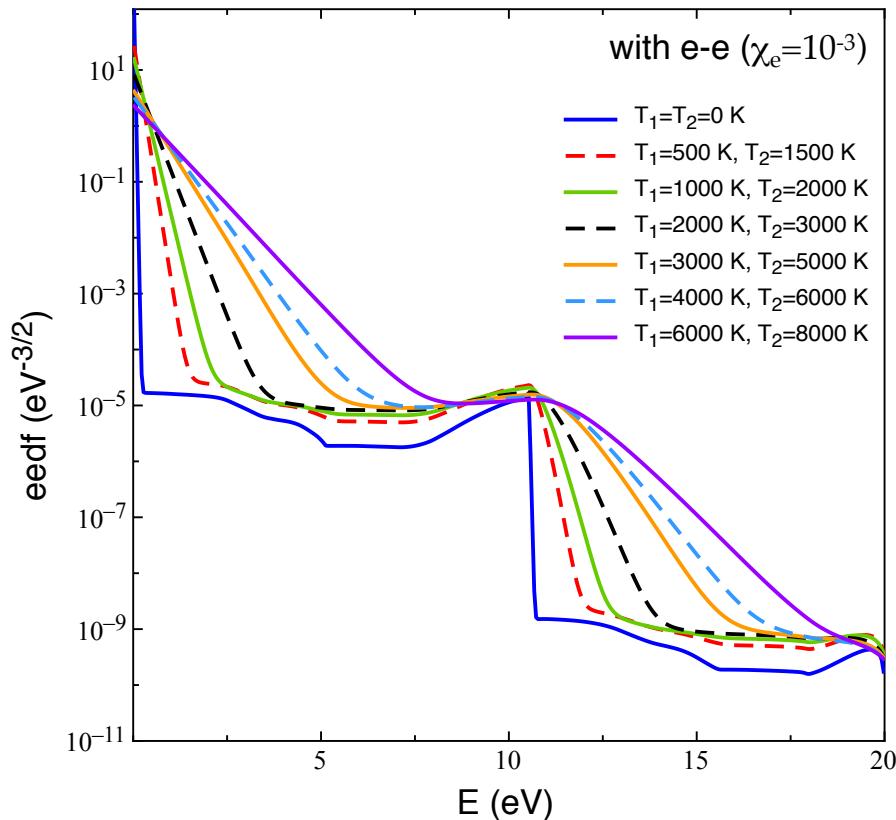


# $\text{CO}_2$ Post Discharge: **Stationary EEDF**

$E/N=0 \text{ Td}$ ,  $\text{CO}_2(e_2)=10^{-4}$

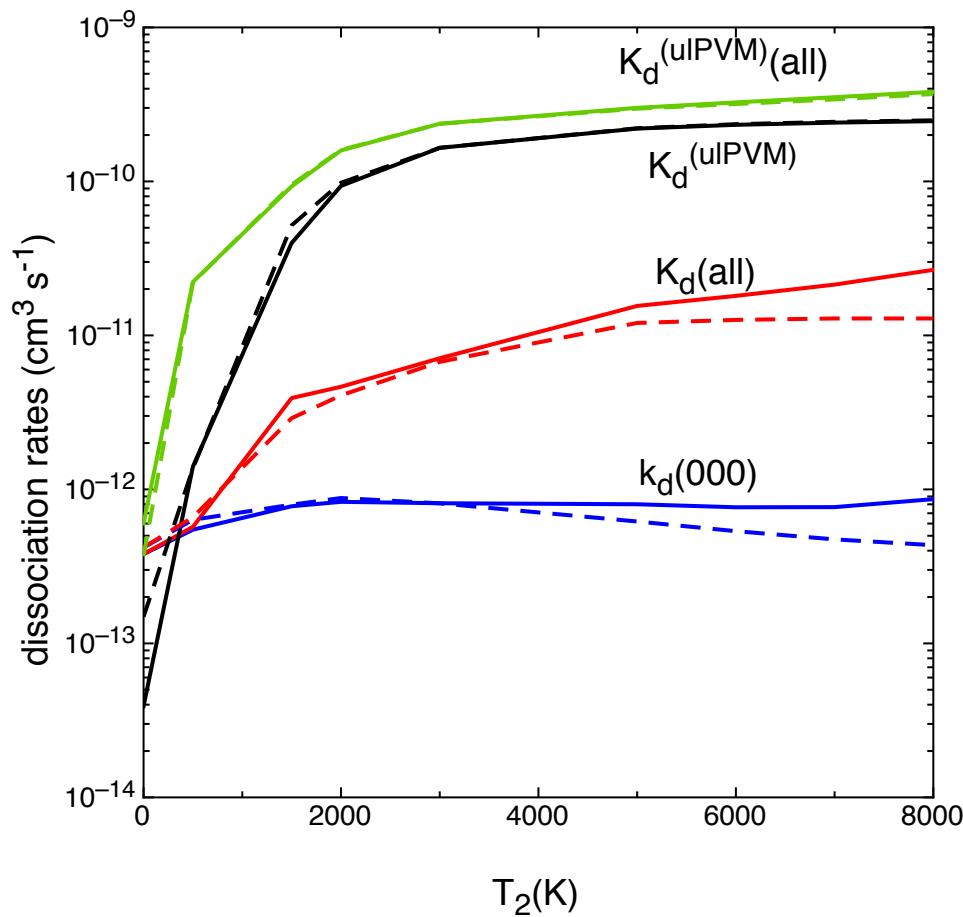
$T_1$  symmetric, bending levels  
 $T_2$  first asymmetric level (001)

## Electron Energy Distribution Function

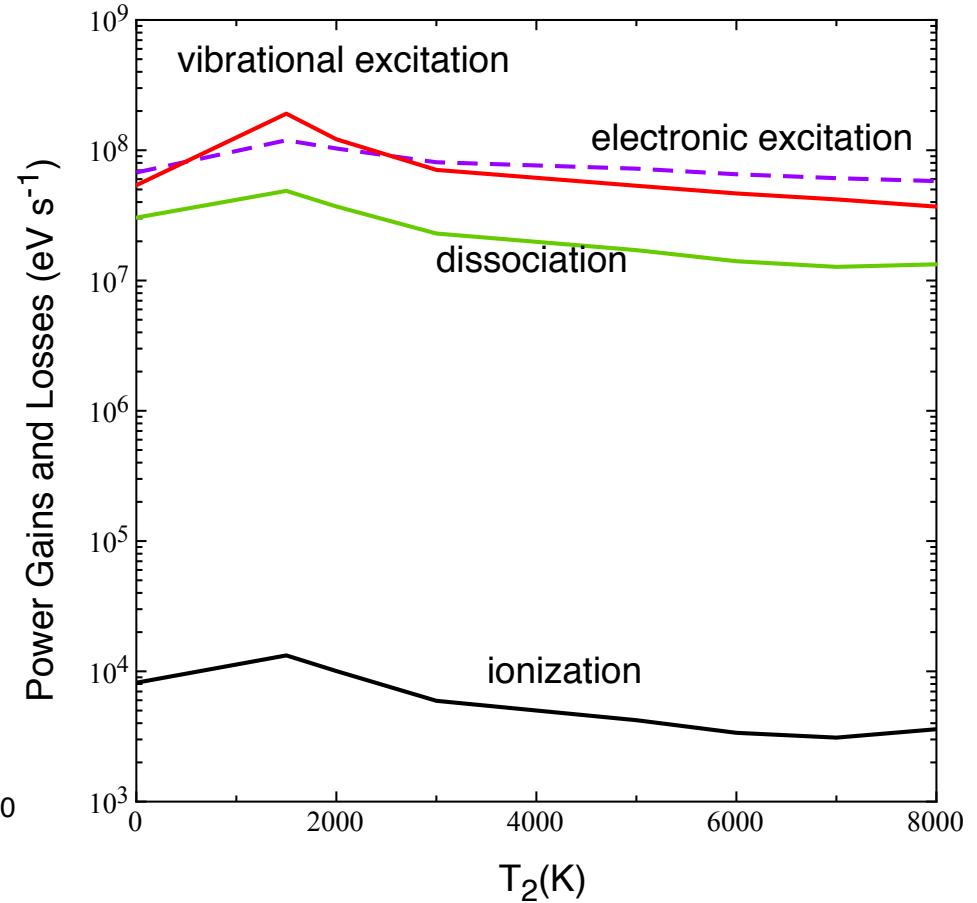


# CO<sub>2</sub> Post Discharge

Dissociation rates

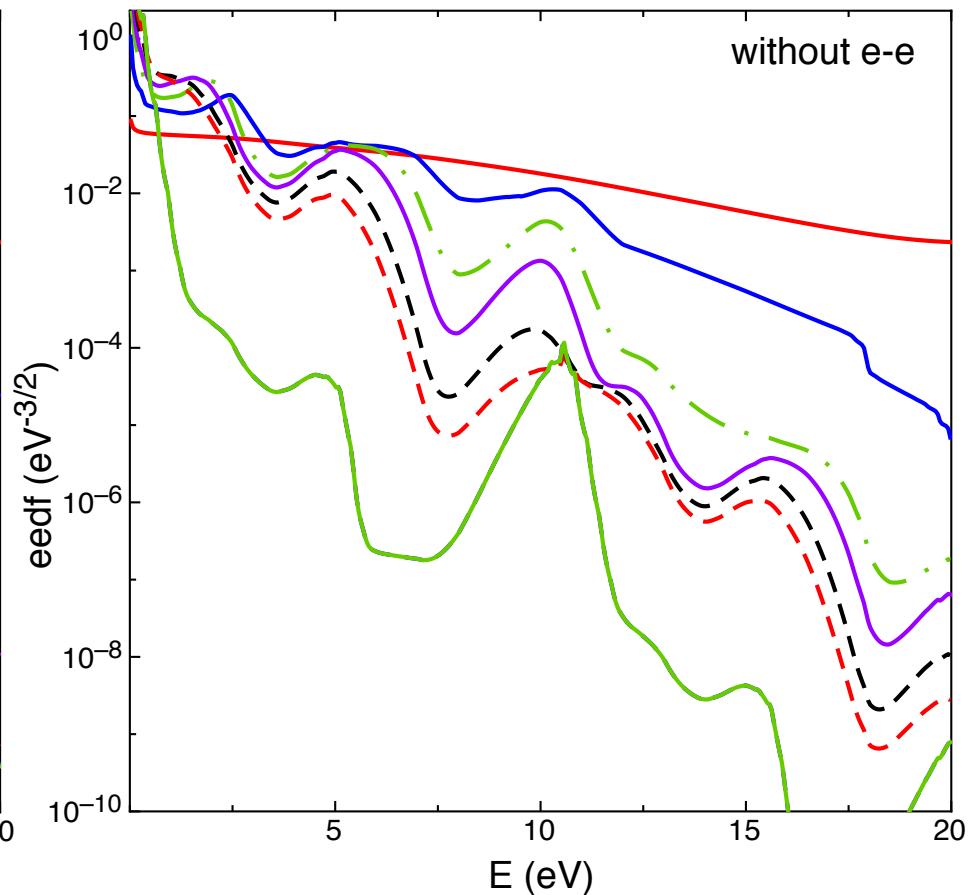
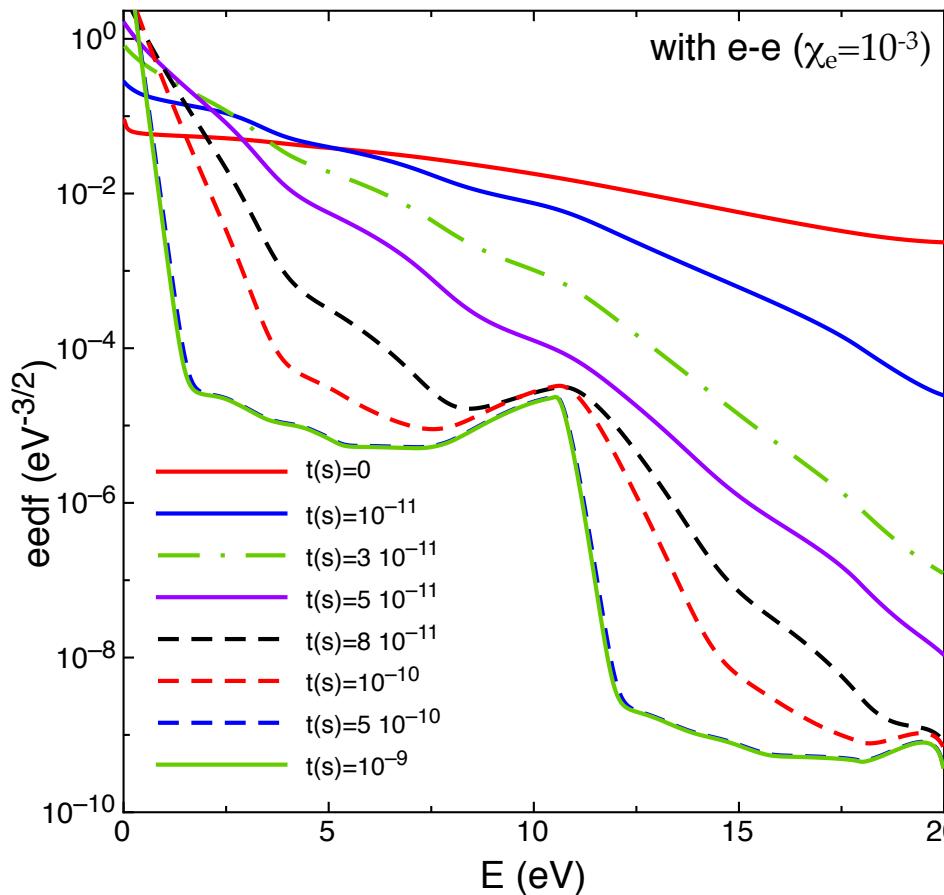


Energy channels



# CO<sub>2</sub> Post Discharge: Time-dependent EEDF

T<sub>1</sub>=500 K, T<sub>2</sub>=1500 K



# E-impact rates dependence on non equilibrium plasma conditions

## Global models

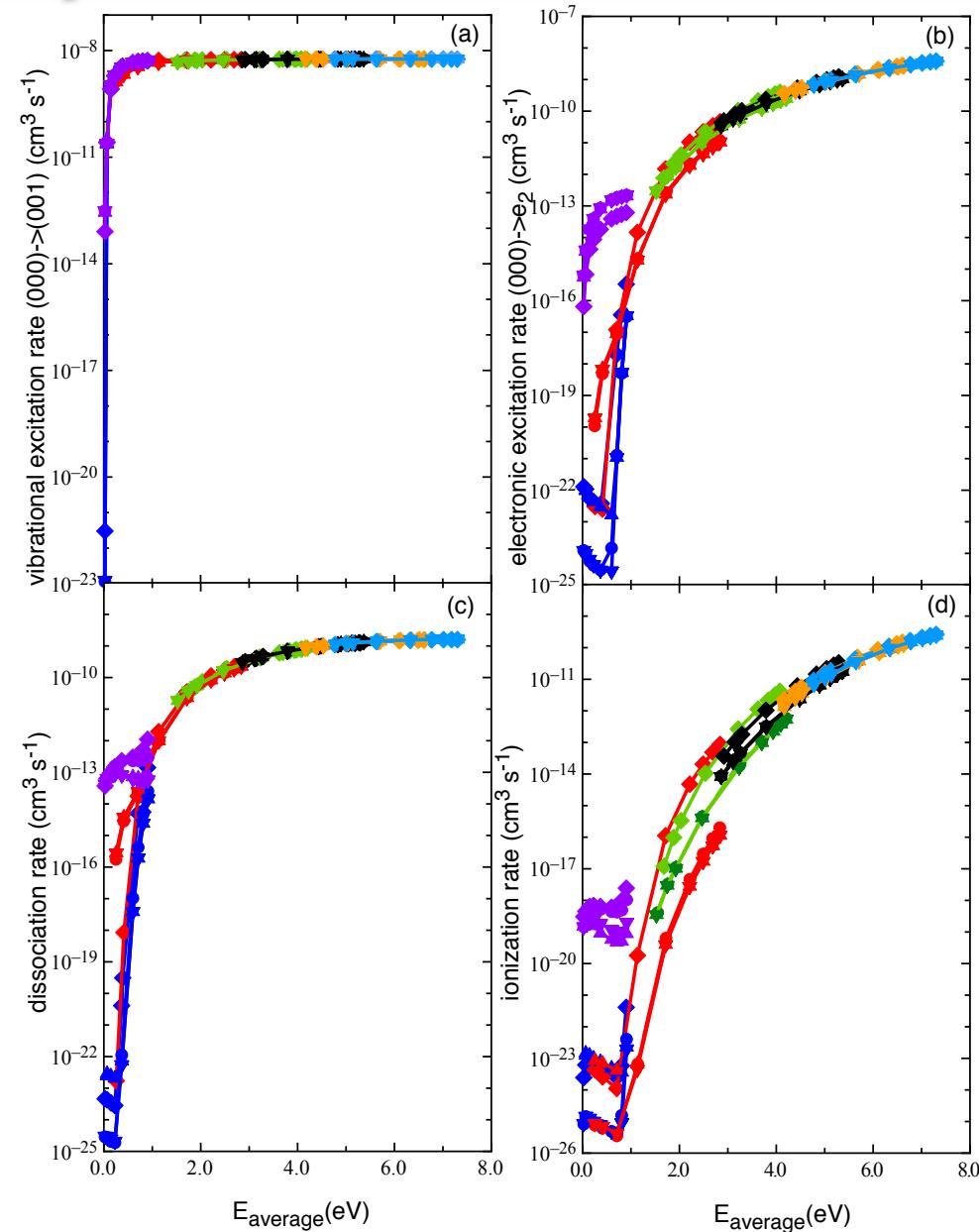
Macroscopic equation for  $T_e$  or  $E_{\text{average}}$

eM rates are calculated by using

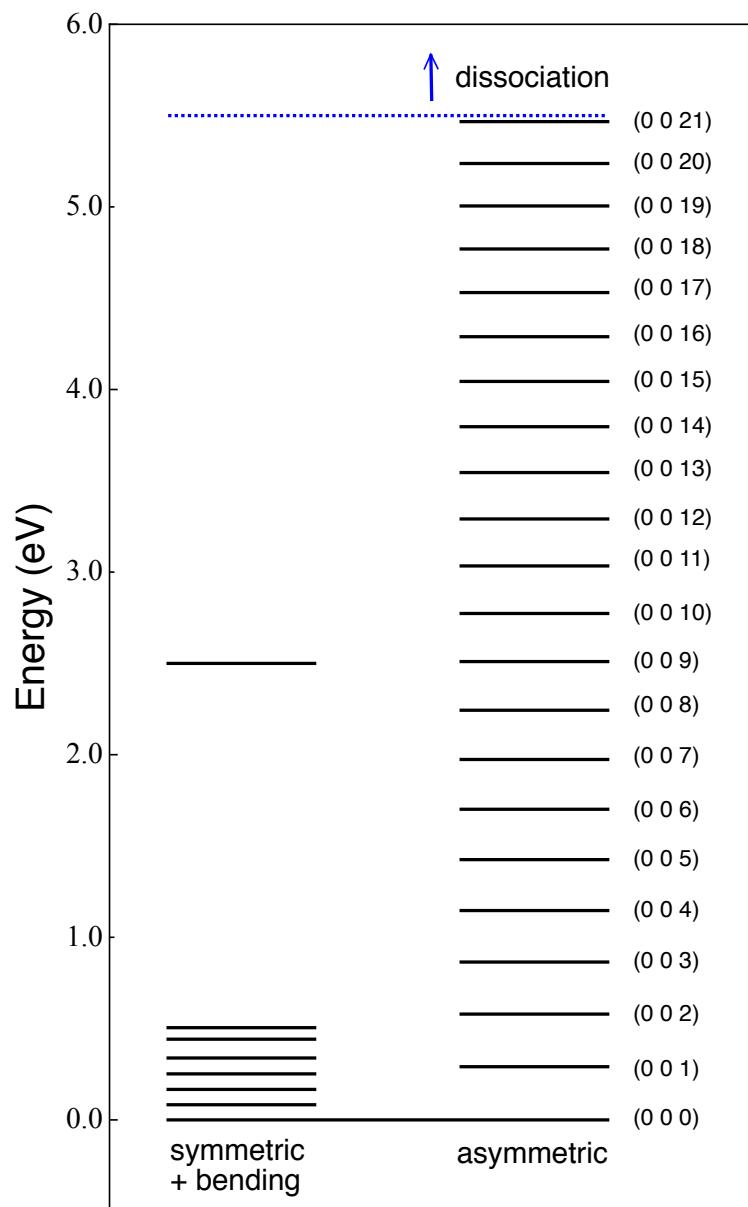
- Maxwell distribution at  $T_e$
- Boltzmann solver (rate vs  $E/N$ )

## Data Collection

- $E/N$  (0, 15, 30, 50, 80, 100 Td)
- Electron molar fraction  $10^{-3}, 10^{-5}$
- with and without e-e collisions



# New set of eV cross sections



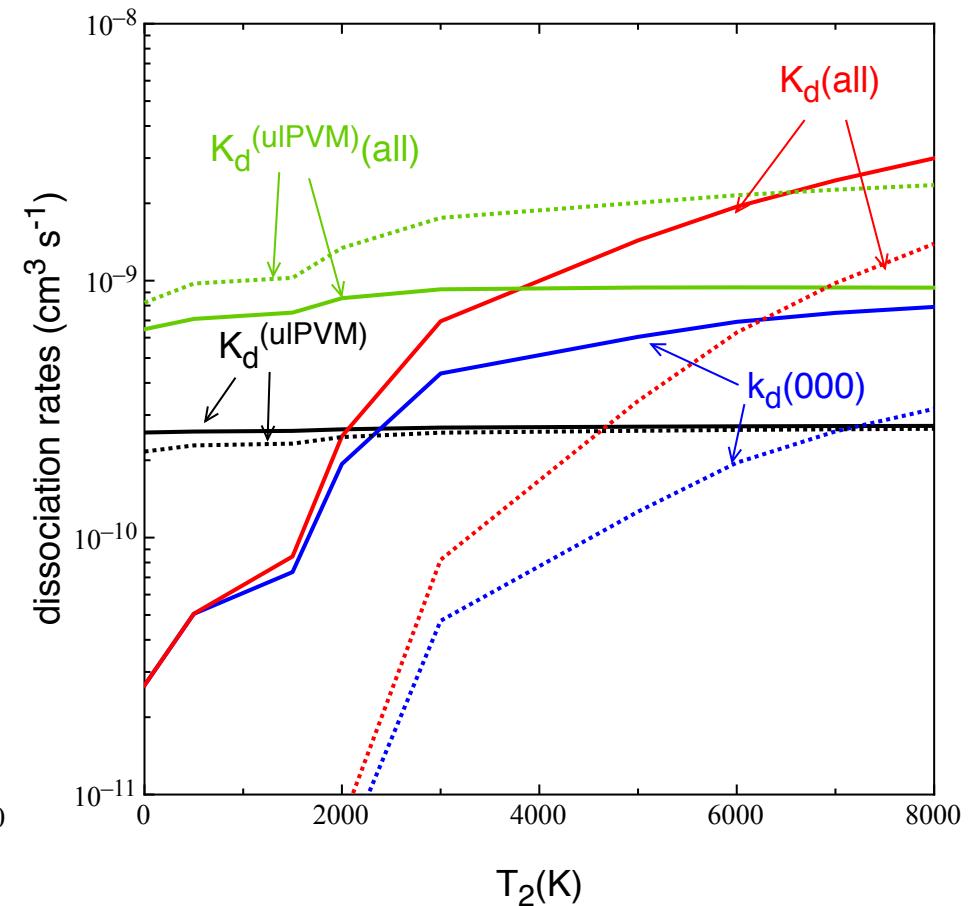
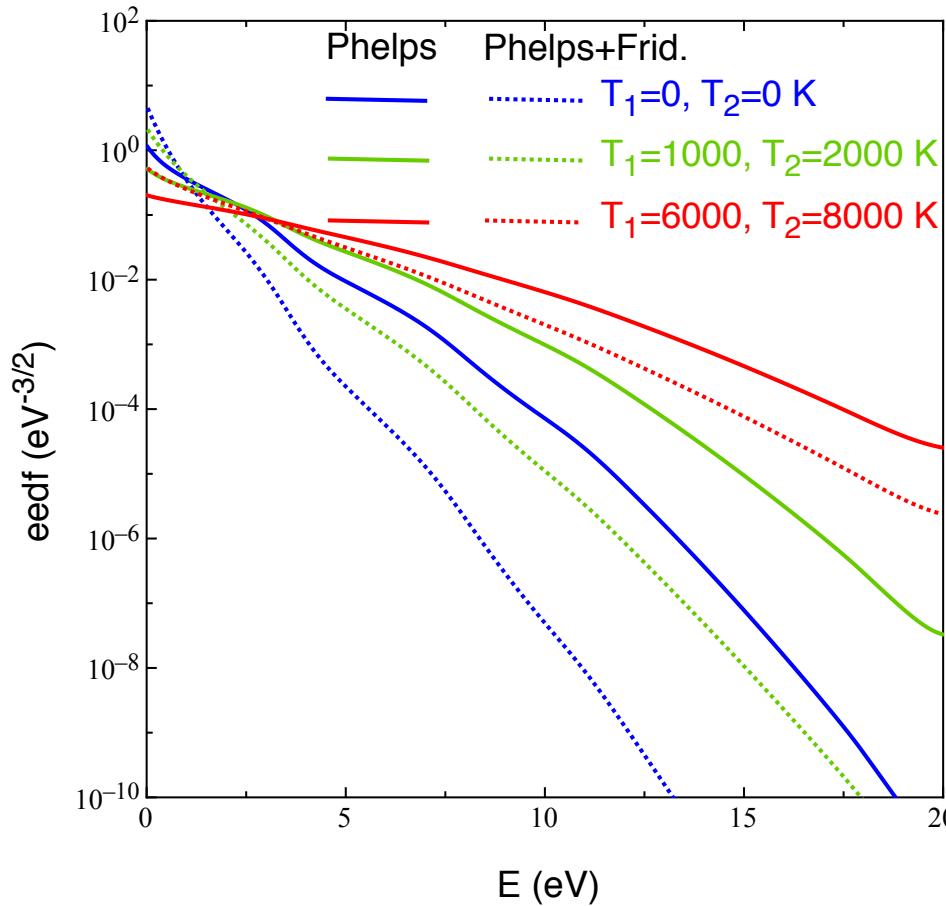
Phelps database



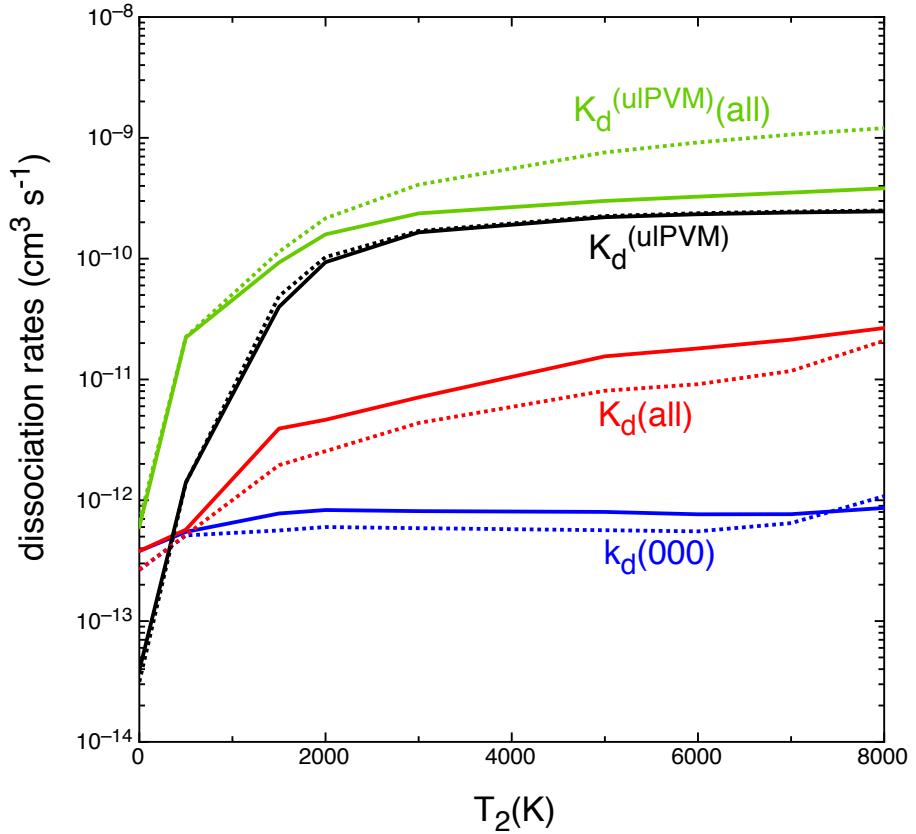
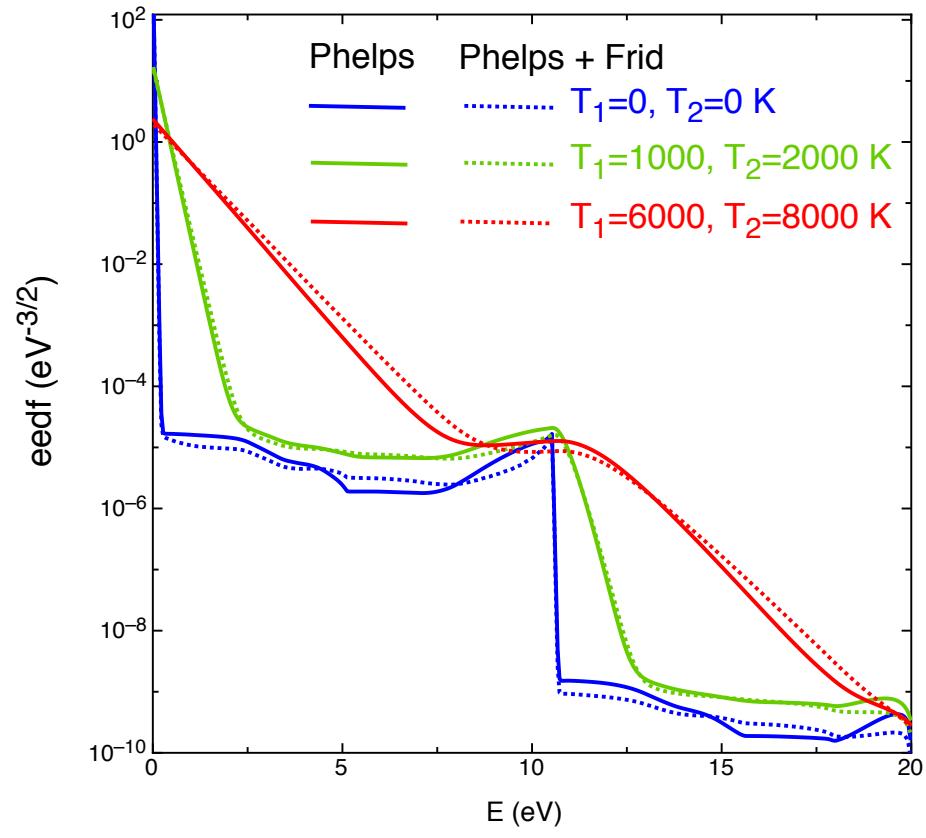
Fridman's semi empirical equation

$$\sigma_{nm}(\varepsilon) = \frac{\exp(-\alpha(m-n-1))}{1+\beta n} \sigma_{01}(\varepsilon + E_{01} - E_{nm})$$

# New set of eV cross sections: continuous discharge



# New set of eV cross sections: post discharge



# Conclusions

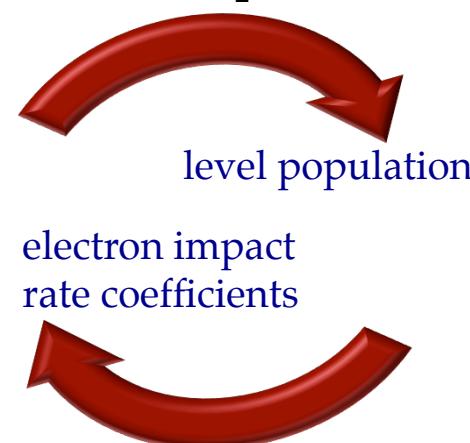
- A parametric solution of the electron Boltzmann equation ( $E/N$ ,  $T_1/T_2$ ,  $c_e$ ,  $\text{CO}_2(e_2)$ ) has been performed in  $\text{CO}_2$  plasma in **continuous discharge** and **post-discharge** conditions.
- Superelastic vibrational and electronic collisions play an important role in affecting the **eefdf**, **dissociation rates** and **electron fractional power losses**.
- **Electron power losses** depend also on **vibrational temperatures**.
- A comparison between **upper limit** dissociation rates of **Pure Vibrational Mechanism** and **electron impact dissociation rates** have been performed
- **Electron impact rates** depend on **non-equilibrium** plasma conditions
- The **accuracy** of the present results largely depends on the eV cross sections used in the Boltzmann solver.

## Future improvements

- A more realistic  $\text{CO}_2$  vibrational ladder (vibrational levels up to the dissociation limit).
- Complete set of VV and VT rates and eV cross sections for such levels.
- **Self consistent coupling** of **eefdf** and  **$\text{CO}_2$  vibrational level kinetics**.

## CHEMICAL KINETICS

- Vibrational kinetics
- Electronic kinetics
- Dissociation/Ionization
- Chemical Composition



## ELECTRON KINETICS